

PACS: 71.15.mb

## ELECTRONIC PROPERTIES OF BULK AND SINGLE-LAYER MoS<sub>2</sub> USING AB INITIO DFT: APPLICATION OF SPIN-ORBIT COUPLING (SOC) PARAMETERS

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Received September 2, 2020; accepted November 4, 2020

Two dimensional (2D) materials are currently gaining a lot of interest due to excellent properties that are different from their bulk structures. Single and few-layered of Transition metal dichalcogenides (TMDCs) have a bandgap that ranges between 1-2 eV, which is used for FET devices or any optoelectronic devices. Within TMDCs, a ton of consideration is focused on Molybdenum Disulfide (MoS<sub>2</sub>) because of its promising band gap-tuning and transition between direct to indirect bandgap properties relies upon its thickness. The density functional theory (DFT) calculations with different functionals and spin-orbit coupling (SOC) parameters were carried out to study the electronic properties of bulk and monolayer MoS<sub>2</sub>. The addition of SOC brought about a noteworthy change in the profile of the band energy, explicitly the splitting of the valence band maximum (VBM) into two sub-bands. The indirect bandgap in bulk MoS<sub>2</sub> ranges from 1.17-1.71 eV and that of the monolayer bandgap was 1.6-1.71 eV. The calculated parameters were compared to the obtained experimental and theoretical results. The obtained density of states (DOS) can be used in explaining the nature of bandgap in both the bulk and monolayer MoS<sub>2</sub>. These electronic characteristics are important for applications in material devices and energy-saving applications.

**KEYWORDS:** Electronic properties, Density functional theory, Spin-orbit coupling, Density of states, MoS<sub>2</sub>, bandgap

One of the greatest discoveries in science is the successful flaking of graphene into a 2D sheet of carbon in 2004 by Novoselov et al [1]. Graphene exhibits excellent mechanical, electronic, and physical properties. It's unanticipated Zero bandgap limits its applications [2, 3]. The introduction of graphene has resulted in a great interest in other 2D materials such as phosphorene [4], silicone [5], TMDCs [6], and monochalcogenides [7, 8], creating huge attention in Nanoelectronic and optoelectronic devices [9-11]. From the various available 2D materials, 2D TMDCs exhibits superb physical properties like large bandgap which is needed in Photocatalytic water splitting [12,13].

Layered TMDCs materials are the best candidates for 2D materials because their bandgap lies between 1 to 2 eV [14]. Within the TMDCs, MoS<sub>2</sub> is the most sought for due to its industrial application starting from the use in lubricant [15], energy storage [16], photovoltaics [17], and catalyst [18]. Bulk MoS<sub>2</sub> is a semiconductor having an indirect bandgap of 1.23 eV, with its monolayer possessing a direct energy bandgap of 1.8 eV [19]. The electronic structure undergoes an excellent transition upon exfoliation from the bulk [1, 19-23]. Currently, field-effect transistor (FET) based on single-layer MoS<sub>2</sub> using HfO<sub>2</sub> as a gate insulator has been experimentally implemented [21]. These extraordinary properties have made monolayer MoS<sub>2</sub> an interesting material in optoelectronic devices and next-generation FET. Also, the transition from indirect to direct bandgap upon thinning shows excellent photoluminescence in 2D monolayer [24]. Albeit various brilliant investigations have been performed on the basic and electronic properties of MoS<sub>2</sub> [25-28]. The size-subordinate tunability of the electronic properties makes MoS<sub>2</sub> a novel material for nanoscale field-effect transistors and optical sensors [29-31]. However, as far as we could possibly know, the impact of SOC on both bulk and monolayer MoS<sub>2</sub> band structure has not been widely examined.

In this study, the band structure of bulk and monolayer MoS<sub>2</sub> were numerically investigated by the plane-wave based CASTEP, and local density approximation (LDA) and generalized gradient approximation (GGA) exchange-correlation functional were used. The influence of the spin-orbit coupling (SOC) effect on the electronic energy band structure of both monolayer and bulk MoS<sub>2</sub> has also been calculated in this study. Our results indicate that both bulk and monolayer MoS<sub>2</sub> materials are very good candidates for Opto- and spintronic device applications.

### COMPUTATIONAL METHOD

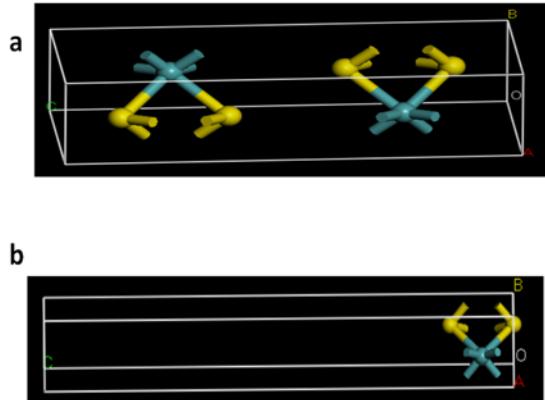
First principle's calculation based on the pseudopotential plane-wave method [32] is executed in the Cambridge serial total energy package (CASTEP) [33]. This method has been used to study the electronic properties of monolayer GeI [34] and bulk and monolayer WSe<sub>2</sub>. The generalized gradient approximations (GGA) of Perdew-Burke-Ernzerhof (PBE) and Local density approximations (LDA) functional [35, 36] including the spin-orbit coupling (SOC) [37] for exchange-correlation interactions. Furthermore, norm-conserving pseudopotential was used in the treatment of ion-electron exchange. The energy cut off for all calculations in this work was 700 eV, for both bulk and monolayer MoS<sub>2</sub>, a k-points separation of 0.015/angstrom and SCF tolerance of  $1.0 \times 10^{-6}$  eV/atom were employed. The electronic minimizer was set to all bands/EDFT. Generally, the density mixing is the proposed choice for electronic minimizer, but to perform the calculation with SOC effect, the required electronic minimizer is set to all bands/EDFT in order to make a meaningful

comparison. The Monkhorst-Pack K-point was set to  $7 \times 7 \times 7$  and  $7 \times 7 \times 1$  for bulk and monolayer respectively. For monolayer MoS<sub>2</sub> case, we created 20 Å vacuum along Z axis in order to prevent any interaction between the layers. The pseudo atomic calculation performed for S:  $3S^2, 3P^4$  and Mo:  $4d^5, 5S^1$ . The simulation of the band structure is performed in high symmetry direction  $\Gamma - M - K - \Gamma$ .

## RESULTS

### Structural parameters

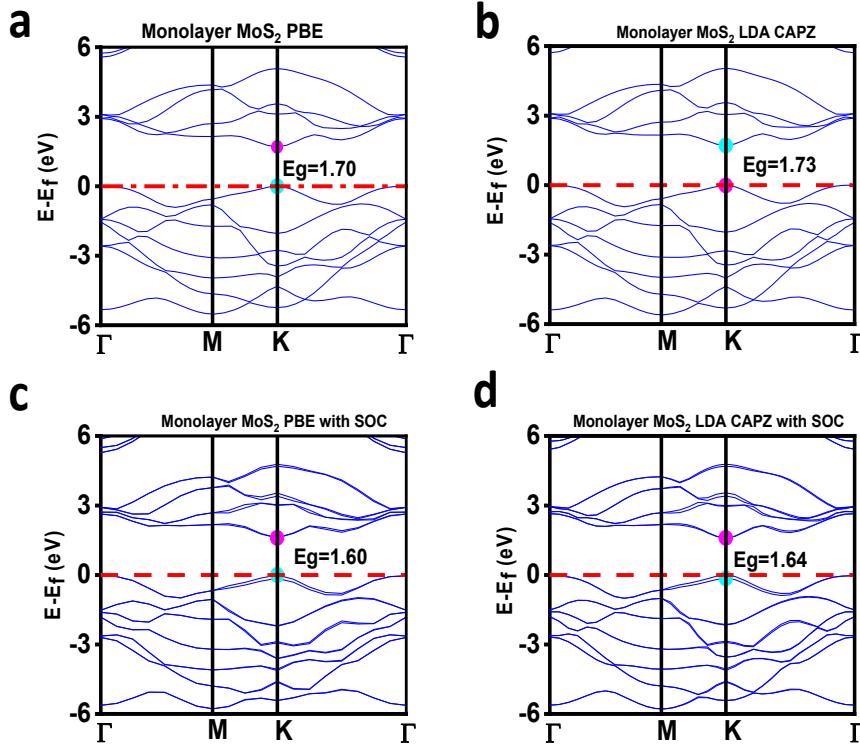
Molybdenum disulfide has a hexagonal crystal structure linked to space group P 6<sub>3</sub>/mmc (194). The structure is made up of the stacking of three atomic layers S-Mo-S layer. Monolayer MoS<sub>2</sub> has a single S-Mo-S layer as shown in Fig. 1b, whiles the bulk MoS<sub>2</sub> has a two-layer monolayer MoS<sub>2</sub> with one-layer Mo atoms place on top of the Sulphur atoms of the other layer, illustrated in Fig. 1b. From geometry optimization, the lattice constants for bulk MoS<sub>2</sub> are 3.192 Å in a and b axis and 12.478 in the c-axis. The monolayer slab was cleaved in 001 direction and a vacuum layer of 20 Å.



**Figure 1.** A ball and stick model of one-unit cell of the (a) bulk MoS<sub>2</sub> and (b) Monolayer MoS<sub>2</sub>. The blue atoms represent the Mo atoms, and the yellow atoms represent the S atoms.

### Electronic properties and density of states

A series of calculations were performed on the bulk and monolayer MoS<sub>2</sub> with LDA CAPZ and GGA PBE functionals incorporated with or without SOC. The electronic band pattern of monolayer MoS<sub>2</sub> as shown in Fig. 2 has a direct bandgap situated at  $K \rightarrow K$  of k point.



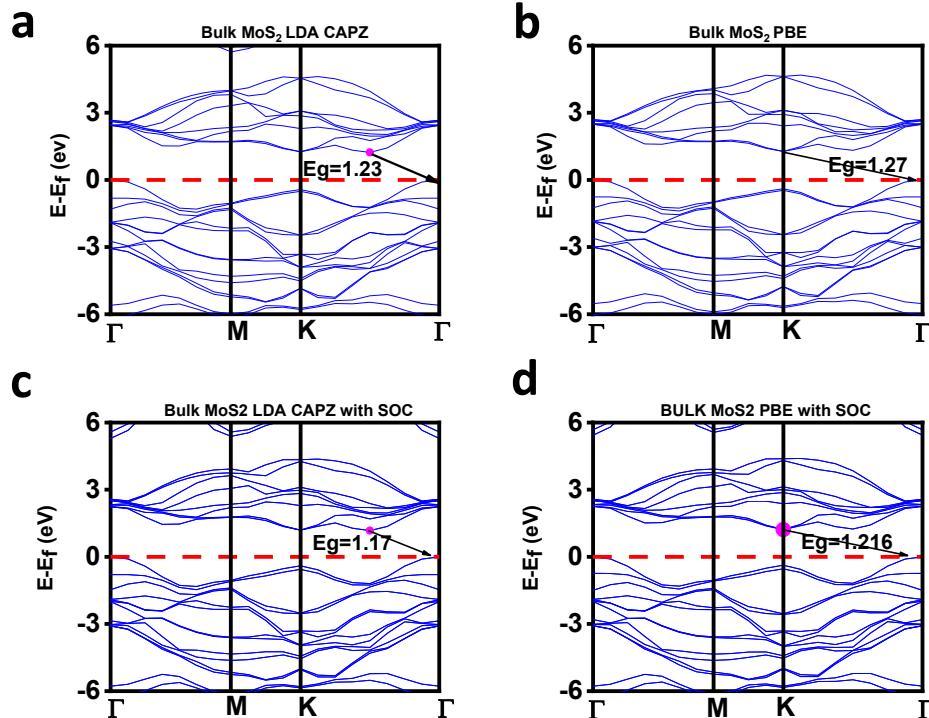
**Figure 2.** Band structure of (a)-monolayer MoS<sub>2</sub> PBE without SOC, (b)-Monolayer LDA CAPZ without SOC, (c)-Monolayer MoS<sub>2</sub> PBE with SOC, (d)-Monolayer MoS<sub>2</sub> LDA CAPZ with SOC

This transition in bandgap makes monolayer MoS<sub>2</sub> applicable to different devices. Interestingly, all the calculations either with or without SOC captures the direct bandgap feature for monolayer MoS<sub>2</sub>. Also, the application of SOC for monolayer MoS<sub>2</sub> affects the band gap value. Furthermore, the valence band maximum and that of the conduction band both occur at high symmetric K point indicating a direct bandgap semiconductor as shown in Fig. 2(a-d) above. The bandgap value for monolayer MoS<sub>2</sub> ranges from 1.60 to 1.71 eV. The experimental result indicates a slight shift from the theoretical result. The theoretical work gives a slight difference in bandgap than the experiment as illustrated in Table.

**Table**  
Energy band gap (eV) and bandgap type for both Bulk and Monolayer

Functionals	Bulk MoS <sub>2</sub>	Bandgap value	Monolayer MoS <sub>2</sub>	Bandgap value
	Indirect bandgap		Direct bandgap	
GGA PBE no SOC	Yes	1.27	Yes	1.70
LDA CAPZ no SOC	Yes	1.23	Yes	1.73
GGA PBE SOC	Yes	1.22	Yes	1.60
LDA CAPZ SOC	Yes	1.17	Yes	1.64
EXP REF	Yes	1.23(ref [19]) 1.29(ref [27])	Yes	1.80 (ref [19])
THEOR REF	Yes	0.7(ref [38]) 1.05(ref [26]) 1.15(ref [39])	Yes	1.55 (ref [25,26]) 1.70 (ref [40]) 1.78 (ref [41]) 1.9 (ref [42])

From Fig. 2(a-d), the valence band maximum (VBM) and the conduction band minimum (CBM) lie below and above the Fermi level respectively along with the K-symmetry. For the monolayer MoS<sub>2</sub>, there is a reasonable band splitting as seen in Fig 2 (c) and (d), this band splitting is brought about by the utilization of SOC and can likewise help in the application of spintronics and valleytronics [43-47]. The band splitting is observed distinctly in the valence band (VB) of the high symmetry K-direction. This result is good agreement with previously reported literature [28, 43, 48].

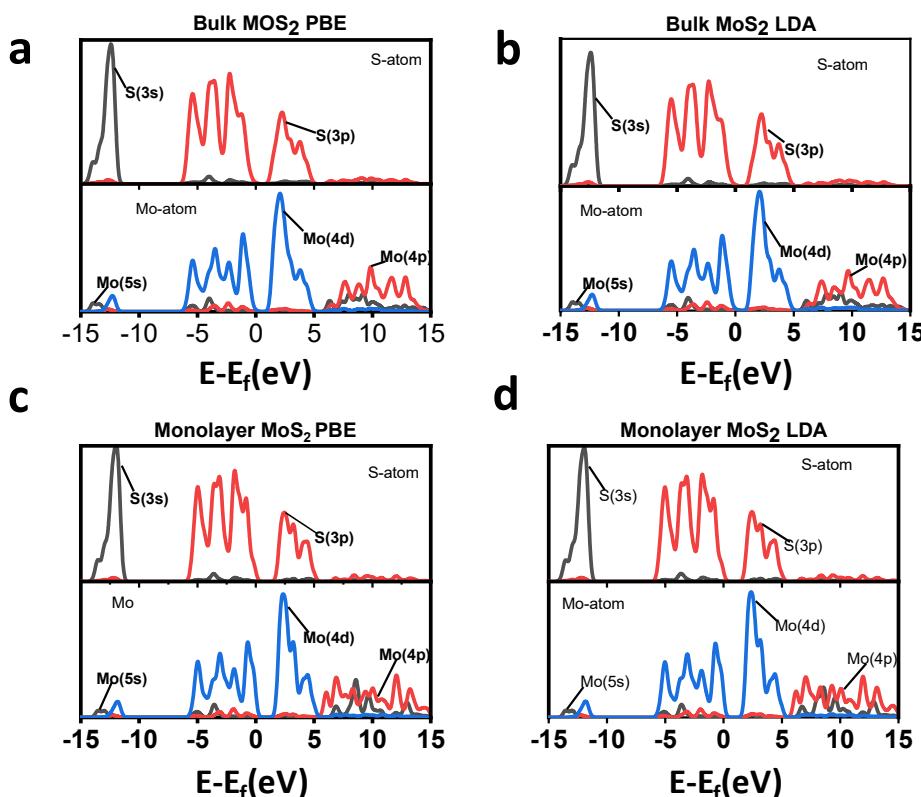


**Figure 3.** Band structure of (a) Bulk MoS<sub>2</sub> PBE without SOC (b) Bulk LDA CAPZ without SOC (c) Bulk MoS<sub>2</sub> PBE with SOC (d) Bulk MoS<sub>2</sub> LDA CAPZ with SOC

Fig. 3. (a)-(d) represents the band structure of bulk MoS<sub>2</sub> for PBE and LDA functionals. The band structure profiles obtained indicate an indirect semiconductor since the maxima of the valence band is situated at KΓ-path whereas that of the minimum of the conduction band is located at Γ-point. The obtained bandgap values range from 1.17-1.27 eV which is consistent with the experimental value [19, 27] shown in Table. From the obtained results of the electronic structures,

we realized that there are slight changes in the energy band with or without SOC. Also, it was observed that the bandgap value of the compound with SOC is smaller than the one without SOC. Fig. 3 (c) and (d) show a slight Valence band (VB) splitting on a high symmetry K direction. This splitting can be ascribed to the utilization of SOC [43, 49-53]. The SOC influences the valence band of the compound as appeared in Fig. 3(c,d).

Fig. 4. (a-d) represents the partial density of states of bulk and monolayer MoS<sub>2</sub> using two different functionals i.e. PBE and LDA. MoS<sub>2</sub> exhibits a neutral charge due to the presence of +4 in the valence of Mo and a charge of -2 for the S- atom. From Fig. 4(a-d), the partial density of states can be divided into three classes of states. For the first class, the density around -14 eV is due to 3s orbital of S atom which is separated by a wide gap. The second class is the one below the Fermi level in which the contributing orbitals the 3p orbitals of S atom and 4d orbitals of Mo. This class possesses a strong hybridization. The third class is the one above the Fermi-level, here the contributing orbital is the 4d orbitals of Mo and is separated by the second class with a narrow gap. For both functionals, the bandgap occurs due to the 4d orbitals of Mo and 3p orbitals of S atoms in monolayer and bulk MoS<sub>2</sub>.



**Figure 4.** Partial density of states of (a) Bulk MoS<sub>2</sub> using PBE functional (b) Bulk MoS<sub>2</sub> using LDA (c) MoS<sub>2</sub> monolayer using PBE functional (d) MoS<sub>2</sub> monolayer using LDA

## CONCLUSION

The electronic properties of bulk and monolayer MoS<sub>2</sub> have been studied using first-principle calculations based on the pseudopotential plane-wave method. Our results show that monolayer MoS<sub>2</sub> is a direct semiconductor whereas bulk MoS<sub>2</sub> has an indirect bandgap. Also, our calculations show that the bandgap of monolayer MoS<sub>2</sub> ranges from 1.6-1.73 eV whereas the bulk is 1.17-1.27 eV which is consistent with the available experimental and theoretical results. The electronic structure and density of states indicate many similarities between the monolayer and bulk MoS<sub>2</sub>. The main contribution to the valence and conduction band are 4d states of Mo and 3p orbitals of s atom in both monolayer and bulk MoS<sub>2</sub>. We have discovered that the incorporation of spin-orbit coupling influences the band structures and the splitting of degenerate valence band occurs on high symmetry K-point.

## Conflict of interest

The authors declare that they have no conflict of interest.

## Funding

This research did not receive any specific grant from funding agencies in the public, commercial, or not-for-profit sectors.

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## ПОЧАТКОВІ ДОСЛІДЖЕННЯ ЕЛЕКТРОННИХ ВЛАСТИВОСТЕЙ ОБ'ЄМНОГО ТА ОДНОШАРОВОГО MoS<sub>2</sub> З ВИКОРИСТАННЯМ ДПФ: ЗАСТОСУВАННЯ ПАРАМЕТРІВ СПІН-ОРБІТАЛЬНОГО ЗВ'ЯЗКУ (SOC)

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Двомірні (2D) матеріали в даний час викликають великий інтерес завдяки чудовим властивостям, що відрізняють їх від об'ємних структур. Одношарові та багатошарові дихалькогеніди переходних металів (TMDC) мають ширину забороненої зони, яка коливається в межах 1-2 еВ, що використовується для пристрій FET або будь-яких оптоелектронних пристрій. У випадку TMDC розгляд зосереджено на дисульфіді молібдену (MoS<sub>2</sub>) через перспективи регулювання забороненої зони, а переход між властивостями прямих та непрямих переходів зони залежить від його товщини. Розрахунки теорії щільності функціонала (DFT) з різними функціоналами та параметрами спін-орбітальної зв'язку (SOC) проводились для вивчення електронних властивостей об'ємного та одношарового MoS<sub>2</sub>. Додавання SOC спричинило помітну зміну профілю енергії зони, явно розділивши максимум валентної зони (VBM) на два діапазони. Зона з непрямими переходами об'ємного MoS<sub>2</sub> коливається в межах 1,17-1,71 еВ, а зона у випадку одного шару становить 1,6-1,71 еВ. Розраховані параметри порівнювали з отриманими експериментальними та теоретичними результатами. Отримана щільність станів (DOS) може бути використана для пояснення природи зони як в об'ємному, так і в одношаровому MoS<sub>2</sub>. Ці електронні характеристики важливі для застосування в матеріальних пристроях та енергозберігаючих застосуваннях.

**КЛЮЧОВІ СЛОВА:** електронні властивості, теорія щільності функціонала, спін-орбітальний зв'язок, щільність станів, MoS<sub>2</sub>, заборонена зона

## ПЕРВОНАЧАЛЬНЫЕ ИССЛЕДОВАНИЯ ЭЛЕКТРОННЫХ СВОЙСТВ ОБЪЕМНОГО И ОДНОСЛОЙНОГО MoS<sub>2</sub> С ИСПОЛЬЗОВАНИЕМ ДПФ: ПРИМЕНЕНИЕ ПАРАМЕТРОВ СПИН-ОРБИТАЛЬНОЙ СВЯЗИ (SOC)

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Двухмерные (2D) материалы в настоящее время вызывают большой интерес благодаря замечательным свойствам, отличающим их от объемных структур. Однослойные и многослойные дихалькогениды переходных металлов (TMDC) имеют ширину запрещенной зоны, которая колеблется в пределах 1-2 эВ, что используется для устройств FET или любых оптоэлектронных устройств. В случае TMDC рассмотрение сосредоточено на дисульфиде молибдена (MoS<sub>2</sub>) из-за перспективы регулирования запрещенной зоны, а переход между свойствами прямых и косвенных переходов зоны зависит от его толщины. Расчеты теории плотности функционала (DFT) с различным функционалом и параметрами спин-орбитальной связи (SOC) проводились для изучения электронных свойств объемного и однослойного MoS<sub>2</sub>. Добавление SOC привело к заметному изменению профиля энергии зоны, явно разделив максимум валентной зоны (VBM) на два диапазона. Зона с косвенным переходом объемного MoS<sub>2</sub> колеблется в пределах 1,17-1,71 эВ, а зона в случае одного слоя составляет 1,6-1,71 эВ. Рассчитанные параметры сравнивались с полученными экспериментальными и теоретическими результатами. Полученная плотность состояний (DOS) может быть использована для объяснения природы зоны как в объемном, так и в однослойном MoS<sub>2</sub>. Эти электронные характеристики важны для применения в материальных устройствах и энергосберегающих приложениях.

**КЛЮЧЕВЫЕ СЛОВА:** электронные свойства, теория плотности функционала, спин-орбитальная связь, плотность состояний, MoS<sub>2</sub>, запрещенная зона